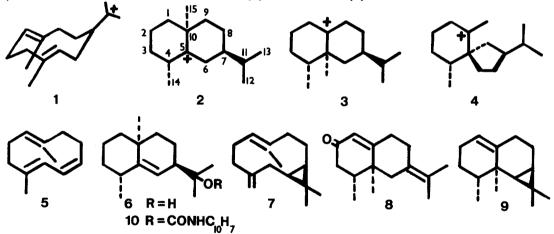
## BIOGENETICALLY SIGNIFICANT SESQUITERPENOIDS FROM RUBUS ROSIFOLIUS OIL<sup>1</sup>

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The importance of cyclodecadienes (1) and 4,10-epi-eudesmanes (2) as precursors in the biogenesis of the nootkatane (3) and spirovetivane (4) sesquiterpenoids has been stressed in recent literature<sup>2-9</sup>. Moreover, the isolation of increasing numbers of epi-eudesmanes from natural sources<sup>5-19</sup> has established this "unusual"<sup>10</sup> configuration as a naturally occurring structural group. This communication reports the isolation of the desisopropyl sesquiterpene pregeijerene (5) and novel isomers of eudesmol and bicyclogermacrene from the steam-volatile leaf oil of the Australian native raspberry *Rubus rosifolius* (Rosaceae). Chemical, spectroscopic and biogenetic evidence suggest that the eudesmol, rosifoliol, has structure (6) and that the bicyclogermacrene is (7). Structure (6) has previously been proposed<sup>2</sup> as a possible precursor of the nootkatones  $\alpha$ -vetivone (8) and calarene (9).



Rosifoliol, the major component (41%) was filtered from the oil of several shrubs collected in the Royal National Park near Sydney. Recrystallisation yielded needles m 84°,  $[\alpha]_D + 105^\circ$ ,  $C_{15}H_{2*}$  (M<sup>+</sup> 204.1876). The presence of OH in the ir (3300 cm<sup>-1</sup>) and pmr ( $\delta$ 1.36) spectra indicated a tertiary alcohol where the molecular ion was not detected<sup>20</sup>. This was confirmed by the formation of a naphthylurethane derivative (10) m 158°,  $[\alpha]_D + 155^\circ$ ,  $C_{26}H_{35}NO_2$  (M<sup>+</sup>391.249) indicating a  $C_{15}H_{26}O$  formula for rosifoliol.

Dehydrogenation of rosifoliol yielded eudalene (11), hydrogenation yielded a dihydroderivative not identical to dihydroeudesmol  $(12)^{21}$  and dehydration yielded (-)- $\delta$ -selinene (13)  $(\left[\alpha\right]_{D} -212^{O}\right)^{11}$ , thus establishing rosifoliol as a 10-epi-eudesmol. Pmr studies with double irradiation and europium shift experiments indicated a selin-5-en-11-ol structure. The spectrum contained one unresolved vinylic doublet at  $\delta$ 5.44 ( $\Delta$ Eu 10.0 ppm/mol.equiv.) showing coupling (J 3.2 Hz) to the proton resonating at  $\delta$ 2.03 ( $\Delta$ Eu 14.8). As the latter signal experienced Eu(dpm)<sub>3</sub> shifts similar in magnitude to the gem-dimethyl signal at  $\delta$ 1.26 ( $\Delta$ Eu 13.7) the vinylic proton can be assigned to the C6 position as in (6). The <sup>13</sup>C nmr proton decoupled and offresonance spectra supported this structure and shifts (Table 1) were assigned using chemical shift theory and published shifts of similar compounds.

Table 1. <sup>13</sup>C Chemical Shifts of Rosifoliol (6).

Carbon position	1*	2	3	4	5	6	7	8	9*	10	11	12 <sup>†</sup>	13 <sup>†</sup>	14 <sup>†</sup>	15
Off-reson. mult.	t	t	t	d	S	d	d	t	t	S .	S	q	q	q	q
δ	39.6	17.7	33.6	38.8	133.2	121.0	45.4	20.3	41.3	34.4	73.5	27.2	27.4	27.8	22.4

\*† cannot distinguish between these resonances.

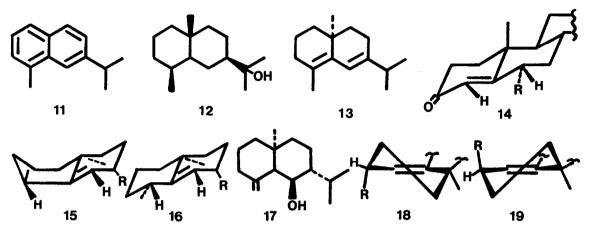
The low  $J_{4,6}$  values obtained for rosifoliol (6) and its naphthylurethane (10) ( $\approx$ 0.0 and 0.5 Hz respectively), indicate, when compared with the analagous  $\Delta^4$ -3-oxo-steriod series  $(14)^{22}$ , a near planar relationship for H4-C4-C5-C6-H6. The W  $h_{/2}$  value of 7 Hz for the C4 proton when the C14 protons are irradiated suggests that the C4 proton lies *gauche* to both of the C3 protons. These conditions are satisfied by two conformers: the C14 $\beta$  methyl boat form (15) and the C14 $\alpha$  methyl chain form (16). As the cyclohexane boat conformer is less stable except when favoured for steric reasons<sup>23</sup> and as the  $4\beta$ ,  $10\alpha$  combination is rare in nature, (16) is favoured.

Some authors<sup>2</sup> would assume a 7 $\beta$  configuration despite the existence of 7 $\alpha$ -substitution (e.g. laevojuneol, (17)<sup>17,18</sup>). The J<sub>6,7</sub> values of 3.2 for rosifoliol (6) and 3.0 Hz for the  $\alpha$ -naphthylurethane (10) indicate a typical cyclohexene system with a dihedral angle of approximately 43<sup>024</sup>. J<sub>7,80</sub> and J<sub>7,80</sub> (4.7 and 6.5 Hz) for (10) indicate that the C7 proton lies *gauche* to both the C8 protons suggesting that the isopropanol group is either *pseudo*-axial  $\alpha$  (18) or *pseudo*-axial  $\beta$  (19). The former is eliminated on the grounds that the bulky  $\alpha$ -substituent is sterically incompatible with the C10  $\alpha$  methyl group (this is not the case with (17) where the substituent group would be equatorial to the fully saturated ring). With rosifoliol, the steric interactions of the saturated system have been relieved<sup>9</sup> by unsaturation at C5.

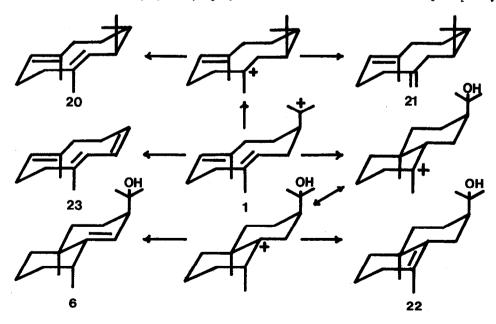
Thus the structure and absolute configuration of rosifoliol can be represented as 4R, 7R, 10S-selin-5-en-11-ol (6) pending confirmation by synthesis.

Pregeijerene (5) constituting 20% of the oil was removed by extraction with aqueous  $AgNO_3^{25}$ . The remaining fraction contained a  $C_{15}H_{24}$  (M<sup>+</sup> 204.1870) hydrocarbon (12% of the oil)  $\left[\alpha\right]_D -13^0$  (crude) showing the presence of a terminal methylene group (887, 3070 cm<sup>-1</sup>,

 $\delta 4.76, 4.88$ ), a trisubstituted double bond ( $\delta 5.25$  1Hbm) with one methyl substituent ( $\delta 1.57$  3Hd J 1Hz), a cyclopropane ring ( $\delta 0.8-0.9$ ) and a *gem*-dimethyl group ( $\delta 0.97$  3Hs, 0.99 3Hs). These results suggest a bicyclogermacra-1(10),4(14)-diene structure (7) which in view of the epimeric nature of rosifoliol (6) and the conformation of bicyclogermacra-1(10),4(5)-diene(20)<sup>26</sup> is most likely to have conformation (21).



The formation of these three *R.* rosifolius constituents and the related bicyclogermacra-1(10),4(5)-diene (20) and 10-epi- $\gamma$ -eudesmol (22)<sup>19</sup> can be accounted for with currently accepted biogenetic theory from carbocation (1) as shown in Scheme 1. Such a pathway would imply a different conformer (23) for pregeijerene than that obtained from *Geijera parviflora*<sup>27</sup>.



Scheme 1

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## **References and Notes**

- Presented at the 10th IUPAC Symposium on the Chemistry of Natural Products, Dunedin, August 1976.
- 2 D.F. MacSweeney, R. Ramage and A. Sattar, Tetrahedron Letters 1970, 557.
- 3 N.H. Anderson and D.D. Syrdal, Tetrahedron Letters 1970, 2277.
- 4 D. Caine and S.L. Graham, Tetrahedron Letters 1976, 2521.
- 5 W. Parker, J.S. Roberts and R. Ramage, Quart. Rev. 1967, 21, 331.
- 6 N.H. Anderson, M.S. Falcone and D.D. Syrdal, Tetrahedron Letters 1970, 1759.
- 7 N.H. Anderson, Phytochemistry 1970, 9, 145.
- 8 J. Garnero, Parf. Cosm. Sav. France 1971, 1, 569.
- 9 J.A. Marshall, St.F. Brady and N.H. Anderson, The Chemistry of Spiro [4,5] Decane Sesquiterpenes, in Chemistry of Organic Natural Products, Ed. W. Herz, H. Griseback, G.W. Kirby, 31, 283, Springer-Verlag, New York 1974.
- 10 A.G. Hortmann and J.B. De Roos, J. Org. Chem. 1969, 34, 736.
- 11 M.L. Maheshwari, T.C. Jain, R.B. Bates, S.C. Bhattacharyya Tetrahedron 1963, 19, 1079.
- 12 A.F. Thomas and M. Ozainne, Tetrahedron Letters 1976, 1717 and assoc. refs.
- 13 H. Schildnecht, H. Holtkotte, D. Krauβ and H. Tacheci, Justus Liebigs Ann. Chem. 1975, 1850.
- 14 L.J. Wadhams, R. Baker and P.E. Howse, Tetrahedron Letters 1974, 1697.
- 15 J.W. Huffmann and L.H. Zalkow, Tetrahedron Letters 1973, 751 and assoc. refs.
- 16 A. Stoessi, E.W.B. Ward and J.B. Stothers, *Tetrahedron Letters*, 1976, 3271 and assoc. refs.
- 17 A.J. Weinheimer, W.W. Youngblood, P.H. Washecheck, T.K.B. Karns and L.S. Ciereszko, *Tetrahedron Letters* 1970, 497.
- 18 (a) S.C. Bhattacharyya, A.S. Rao and A.M. Shaligramm, *Chem. Ind.* 1960, 469. (b) R. Seshadri, P.S. Kalsi, K.K. Chakravarti and S.C. Bhattacharyya, *Tetrahedron* 1967, 23, 1269.
- 19 R. Kaiser and P. Naegeli, Tetrahedron Letters 1972, 2009.
- 20 R.A. Friedel, J.L. Schultz and A.G. Sharkey, Anal. Chem. 1956, 28, 926.
- 21 F.J. McQuillin and J.D. Parrack, J. Chem. Soc. 1956, 2973.
- 22 D.J. Collins, J.J. Hobbs and S. Sternhell, Aust. J. Chem. 1963, 16, 1030.
- 23 (a) D.T. Cropp, B.B. Deshurst and J.S.E. Holker, Chem. Ind. 1961, 209. (b) B.B. Dewhurst, J.S.E. Holker, A. Lablache-Combier, J. Levisalles, Chem. Ind., 1961, 1667.
- 24 G.V. Smith and H. Kriloff, J. Amer. Chem. Soc. 1963, 85, 2016.
- 25 R.V.H. Jones and M.D. Sutherland, Aust. J. Chem., 1968, 21, 2255.
- 26 K. Takeda, I. Horibe and H. Minato, J.C.S. Chem. Comm., 1971, 308.
- 27 R.J. McClure, G.I. Sim, P. Coggon and A.T. McPhail, J.C.S. Chem. Comm. 1970, 128.